

FORMULATION AND EVALUATION OF FLOATIN GBEADSO OF OF LOXACINUSINGELECTRONMICROSCOPY

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ABSTRACT: While in mid-air Pharmacological delivery systems aim to increase the amount of time a drug spends in the stomach after oral ingestion. Many doctors recommend the fluoroquinolone antibiotic ofloxacin for the treatment of stomach and duodenal ulcers. Its solubility varies with pH, being somewhat soluble at neutral and somewhat insoluble at acidic pH. Adverse effects on absorption in the lower parts of the gut arise due to drug precipitation in the intestine. Therefore, systems that remain in the stomach for an extended length of time and release the active ingredient gradually are required. The current research set out to find a way to distribute ofloxacin in a way that would allow it to be retained for developing ofloxacin-containing rice bran oil-entrapped zinc pectinate beads, which increases local activity in the stomach area against *Helicobacter pylori*. Different types of flocculent ofloxacin beads were created by combining Low Methoxy Pectin with other polymers that influence the rate of flocculation, such as Xanthan Gum, Gellan Gum, or Karaya Gum. The beads were made using the emulsion gelation technique. The beads were given buoyancy by using rice bran oil, which has a low density. The spherical beads exhibited the maximum drug content and drug trapping when made with a combination of Low Methoxy Pectin and Gellan Gum. To achieve the desired level of buoyancy, the beads had to be treated with at least 25% w/w rice bran oil, according to the buoyancy tests.

Keywords: Antibiotic, Absorption window, Gastric residence time, Polymer mixture, Stomach

INTRODUCTION: Several physiological constraints are present in most oral dose forms. These include uneven drug release, shorter residence time in the stomach, and variable gastrointestinal transit due to varying gastric emptying.

Because only portion of the medicine is absorbed once it travels down the absorption site, this causes partial absorption of medications with an absorption window in the upper section of the small intestine. Thanks to floating drug delivery systems (FDDS), all of the aforementioned conditions may be satisfied, and the medications can be effectively delivered to the absorption window for local action for the treatment of gastrointestinal problems such as gastro-esophageal reflux. The hydro-dynamically balanced systems (HBS) or floating drug delivery systems (FDDS) are able to float in the stomach for an extended length of time because their bulk density is lower than that of gastric fluids. The medication is released from the system at a controlled pace while the systems float on the stomach contents. Gastric residence time (GRT) increases and, in certain situations, variations in plasma drug concentrations are better controlled as a consequence of this. Efficacious against both Gram-positive and Gram-negative bacteria, ofloxacin is a broad-spectrum antibiotic. An efficient method for treating gastric and duodenal ulcers, as well as eliminating *Helicobacter pylori* from the stomach, is to formulate FDDS of H₂ antagonists.

DATA AND PROCEDURES:

Apex Formulations Pvt. Ltd. of Ahmedabad, Gujarat, India provided the ofloxacin, Sigma Aldrich chemicals of Hyderabad, India provided the gellan gum, and Sigma Aldrich of the United States supplied the Xanthan gum. The rice bran oil and Karaya gum were sourced from Sri Anjanaya Agrotech Pvt. Ltd in Harihar and Mumbai, Karnataka, India, respectively. Analytical grade chemicals and reagents were used in every other case.

Methods:

Development of Calibration Curve: Ofloxacin equivalent to 100 mg was dissolved in 100 ml of 0.1 N hydrochloric acid buffer (pH 1.2) to get a stock solution A. 10 ml was pipetted out and the volume was made up to 100 ml with 0.1 N hydrochloric acid buffer to get a stock solution B.

From the stock solution B, aliquots were diluted to 10 ml with 0.1 N hydrochloric acid buffer to get 1 to 8 µg/ml solutions and measured at 294.5 nm. (Spectrophotometer UV-1601, Shimadzu, Japan)¹.

Drug Polymer Compatibility Studies:Differential

Scanning

Calorimetry

Using Differential Scanning Calorimetry (DSC-60, Shimadzu, Japan), we tested the compatibility of Ofloxacin with various polymers in a physical combination. When the temperature reached 350 °C, it was maintained at 10 °C/min. Purging was done using nitrogen gas at a rate of 30 ml/min.2. Fourier Transform Infrared Spectroscopy (FTIR): An IR spectrophotometer (Shimadzu, model 840, Japan) was used to capture the spectra of the drug and polymers. Any significant interactions between the two were then investigated. For the aim of comparing things and evaluating the peak pattern, this were done qualitatively.

The flotation zinc pectinate beads were prepared by passing Ofloxacin, LMP, GG, XG, and KG through filter no. 80 in isolation. The ofloxacin was dissolved in distilled water at a concentration of 20% by weight of the dry polymer. The aforesaid dispersion was used to dissolve LMP (3% w/v) on its own, as well as polymer mixes (3% w/v) including LMP and GG, LMP and XG, and LMP and KG in three different ratios. One formulation was specifically made using polymer mixtures (3% w/v) containing GG, XG, KG, and LMP.

TABLE1:FORMULATIONVARIABLESOFVARIOUSOFLOXACINBEADFORMULATIONS

| Formulation code | LMP:GG (3%W/V) | LMP:XG (3%W/V) | LMP:KG (3%W/V) | LMP:GG:XG:KG (3%W/V) | RICE BRANOIL(3 %W/V) |
|------------------|----------------|----------------|----------------|----------------------|----------------------|
| F | 10:0 | 10:0 | 10:0 | - | 15 |
| | 10:0 | 10:0 | 10:0 | - | 20 |
| | 10:0 | 10:0 | 10:0 | - | 25 |
| F1 | 9:1 | - | - | - | 25 |
| F2 | 8:2 | - | - | - | 25 |
| F3 | 7:3 | - | - | - | 25 |
| F4 | - | 9:1 | - | - | 25 |
| F5 | - | 8:2 | - | - | 25 |
| F6 | - | 7:3 | - | - | 25 |
| F7 | - | - | 9:1 | - | 25 |
| F8 | - | - | 8:2 | - | 25 |
| F9 | - | - | - | 8:0.66:0.66:0.66 | 25 |

To the above mixture rice bran oil (25% w/w) was added and stirred to form a homogeneous emulsion. The drug-loaded emulsion was extruded through a 23 G syringe needle into zinc chloride solution (5% w/v) maintained under gentle agitation.

The beads were allowed to remain in the same solution for 30 min to improve their mechanical strength. The formed beads were separated, washed with water and allowed to dry at room temperature overnight.

Table 1 lists the formulation variables for different formulations of Ofloxacin loaded floating beads. Blank beads without Ofloxacin were also prepared using the same technique^{3,4}.

Evaluation of Physicochemical Parameters of Floating Beads of Zinc Pectinate:

Determination of Bead Diameter: The diameter of a sample of gel beads (25 beads) of each formulation was determined using a dial thickness meter.

Drug Content: An accurately weighed sample of beads (100mg) was crushed in a mortar and added to 100ml of 0.1N hydrochloric acid buffer (pH 1.2) and kept overnight under stirring to elute completed drug from the polymer matrix. It was filtered and analyzed at 294.5nm (UV spectrophotometer, 1601, Shimadzu, Japan) against blank bead mixture, which was treated similarly. The drug content of each formulation was recorded as mg/100mg of gel beads⁵.

Drug Entrapment Efficiency: The percentage drug entrapment efficiency (% EE) of each bead formulation was calculated using the following equation:⁵

$$EE(\%) = \frac{\text{Actual Drug Content}}{\text{Theoretical Drug Content}} \times 100$$

Determination of Swelling Index: The swelling behavior of the zinc pectinate beads was studied in

0.1 N HCl (pH 1.2) buffer. Approximately 100mg of beads were taken in a dissolution basket and weighed (W_1); the baskets along with the beads were immersed in 0.1N HCl buffer. The weight (W_2) of the basket along with the beads was determined for 8 h: every 30 minutes for the first 2h, and then every h after that. The swelling index (SI) of each formulation was calculated using the following equation:

$$\%SI = \frac{W_2 - W_1}{W_1} \times 100$$

Buoyancy Studies: The time between the introduction of the FDDS into the medium and its buoyancy to the upper one third of the dissolution vessel (floating lag time) and the time for which the formulation constantly floated on the surface of the medium (floating duration) were measured simultaneously as a part of dissolution studies by visual observation⁶.

Invitro Drug Release Studies: *Invitro* release characteristics of ofloxacin floating gel beads (n =3) were evaluated employing USP XIV dissolution testing apparatus 2 (paddle method) using 500ml of 0.1 N HCl buffer as dissolution medium maintained at 37 ± 0.5 °C. The contents were stirred at 50 rpm. A 5ml aliquot of the solution was withdrawn at predetermined time intervals for 8 h and fresh 5ml dissolution media was replaced to maintain sink condition. The sample aliquots were analyzed at a wavelength of 294.5 nm⁷.

Stability Studies: Stability studies were carried out according to ICH guidelines by storing the formulation F1 at 40 ± 2 °C and relative humidity $75 \pm 5\%$ for a period of two months in a programmable environmental test chamber (CHM-10S, Remi Instruments Ltd., Mumbai, India). The samples were withdrawn at 30 and 60 days and analyzed for the drug content, floating behavior and *in vitro* drug release⁸.

Scanning Electron Microscopy (SEM): Morphological examination of the surface and external structure of the dried beads of formulation F1, F4 and F7 (Both drug loaded and blank beads) was performed using a scanning electron microscope (SEM) (model JEOL, JSM-840A). The samples were gold coated prior to the scanning⁹.

RESULTS AND DISCUSSION:

Development of Calibration Curve: Concentration and absorbance obtained for standard plot of Ofloxacin in 0.1 N hydrochloric acid buffer (pH 1.2) are shown in **Fig. 1**.

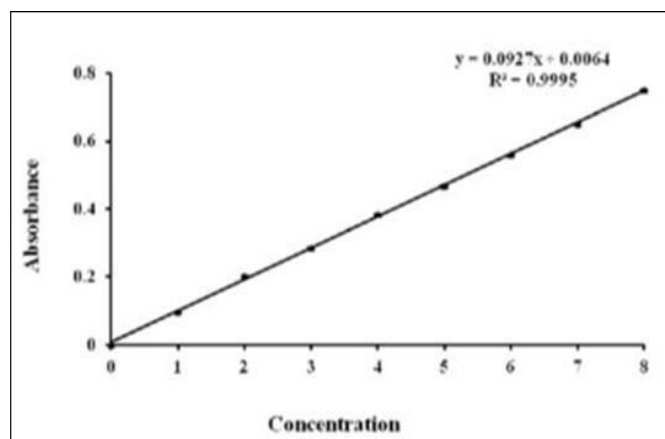


FIG.1: CALIBRATION CURVE OF OFLOXACIN

Drug Polymer Compatibility

Studies: Differential Scanning Calorimetry (DSC): The DSC thermograms of physical mixture of Ofloxacin and the polymers showed that characteristic peaks of polymers and Ofloxacin peaks were still present in the physical mixture but slightly shifted from their original positions as shown in Fig.2. The findings indicate that the drug and polymers are compatible with each other.

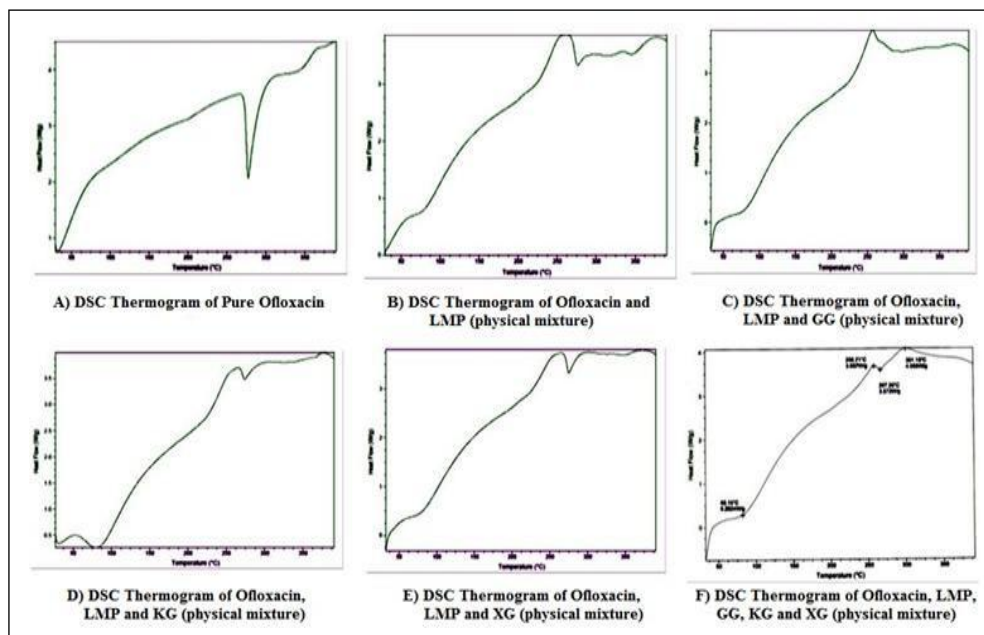


FIG.2: DSC THERMOGRAM OF PURE OFLOXACIN, OFLOXACIN AND LMP, OFLOXACIN, LMP AND GG, OFLOXACIN, LMP AND KG, OFLOXACIN, LMP AND XG, OFLOXACIN, LMP, GG, KG AND XG

Fourier Transform Infrared Radiation (FTIR): All the above bands associated with the pure drug are present in the FTIR spectra of drug in combination with gellangum, karayagum and xanthangum as seen in Fig.3. This shows that there is no chemical interaction taking place between drug and excipients.

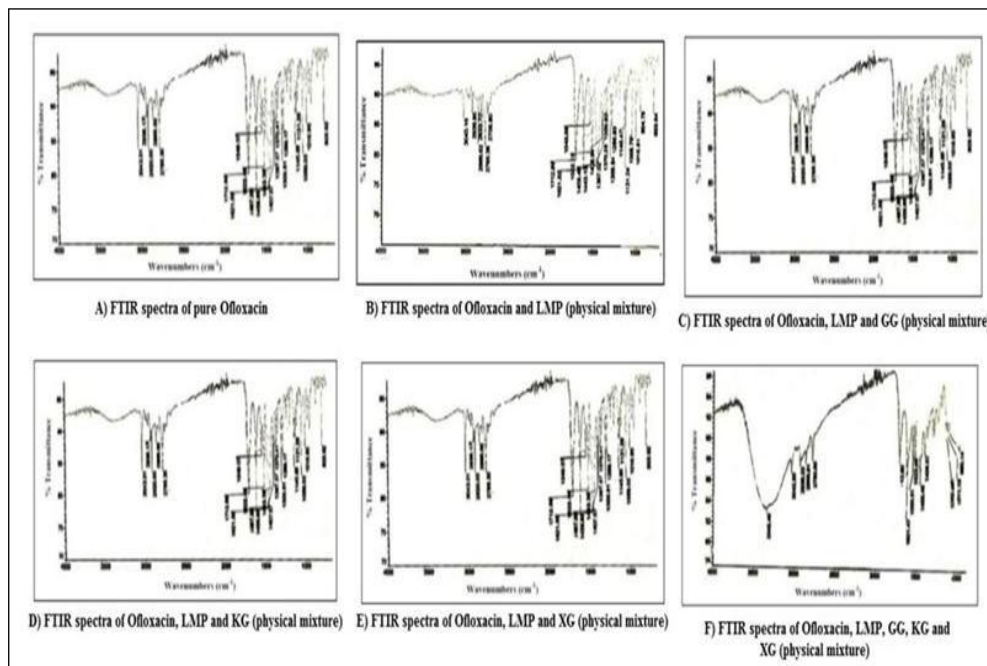


FIG.3: FTIR SPECTRA OF PURE OFLOXACIN, OFLOXACIN AND LMP, OFLOXACIN, LMP AND GG, OFLOXACIN, LMP AND KG, OFLOXACIN, LMP AND XG, OFLOXACIN, LMP, GG, KG AND XG

Evaluation of Physicochemical Parameters of Floating Beads of Zinc Pectinate:

Determination of Bead Diameter: The prepared beads were almost spherical and translucent. The mean surface diameter of 10 formulations was between 1.691 ± 0.022 (mean \pm SD) and 2.099 ± 0.041 (mean \pm SD).

Drug Content, Drug Entrapment and Swelling Index: The percent drug content, entrapment efficiency and swelling index for various Ofloxacin floating beads formulations is shown in **Table 2**.

TABLE 2: CHARACTERIZATION OF FLOATING ZINC PECTINATE BEADS

| Formulation Code | Mean diameter \pm SD (mm) | Drug content (mg) | %EE | %Swelling Index |
|------------------|-----------------------------|-------------------|-------|-----------------|
| Fblank | 1.691 ± 0.022 | - | - | - |
| F | 1.693 ± 0.015 | 2.437 ± 0.037 | 57.49 | 0.74 |
| F1 | 1.751 ± 0.023 | 1.236 ± 0.017 | 78.81 | 2.42 |
| F2 | 1.841 ± 0.022 | 1.620 ± 0.054 | 69.16 | 1.79 |
| F3 | 1.898 ± 0.018 | 1.551 ± 0.114 | 72.13 | 2.21 |
| F4 | 2.193 ± 0.017 | 1.681 ± 0.023 | 76.20 | 1.01 |
| F5 | 1.836 ± 0.018 | 1.621 ± 0.063 | 60.61 | 3.75 |
| F6 | 2.057 ± 0.069 | 1.362 ± 0.035 | 74.28 | 1.22 |
| F7 | 2.009 ± 0.027 | 1.713 ± 0.111 | 65.40 | 2.10 |
| F8 | 2.099 ± 0.041 | 1.417 ± 0.027 | 62.98 | 1.74 |
| F9 | 2.008 ± 0.063 | 1.434 ± 0.240 | 58.95 | 0.72 |

Buoyancy Studies: The floating ability of prepared beads was evaluated along with dissolution studies. The beads without oil sank immediately in 0.1 N HCl (pH 1.2), while beads containing sufficient amount of rice bran oil (25%) demonstrated instantaneous and excellent floating properties (**Table 3**).

TABLE 3: BUOYANCY CHARACTERISTICS OF FLOATING BEADS

| Formulation Code | Amount of oil (% w/w) | FLT (min) | Floating Duration (h) |
|------------------|-----------------------|-----------|-----------------------|
| FBlank | - | NF | NF |
| F | 10 | NF | NF |
| F | 15 | NF | NF |
| F | 25 | 0 | 24 |
| F1 | 25 | 0 | 24 |
| F2 | 25 | 0 | 24 |
| F3 | 25 | 0 | 24 |
| F4 | 25 | 0 | 24 |
| F5 | 25 | 0 | 24 |
| F6 | 25 | 0 | 24 |
| F7 | 25 | 0 | 24 |
| F8 | 25 | 0 | 24 |
| F9 | 25 | 0 | 24 |

In vitro Drug Release Studies: In vitro drug release study of Ofloxacin floating beads was carried in 0.1N HCl (pH 1.2), for a period of 8 h. In the 0.1N HCl, the beads exhibited a biphasic release profile as an initial rapid drug release phase (burst effect) followed by a sustained, gradually increasing drug release phase after 1 h extending up to 8 h. Formulation F contained only LMP could not sustain the Ofloxacin release up to 8 h. It released complete drug at the end of 4 h (**Table 4**).

TABLE 4: INVITRO RELEASE CHARACTERISTICS OF FORMULATION F

| Serial Number | Time (h) | SQRT | Log time | Cum. % drug release | Log % drug remaining | Log % drug release |
|---------------|----------|--------|----------|---------------------|----------------------|--------------------|
| 1 | 0 | 0 | - | 0 | 2 | - |
| 2 | 0.5 | 0.7071 | -0.3010 | 65.92 ± 1.71 | 1.5325 | 1.8190 |
| 3 | 1 | 1 | 0 | 74.47 ± 2.20 | 1.4071 | 1.8721 |
| 4 | 1.5 | 1.2247 | 0.1760 | 82.01 ± 0.77 | 1.2550 | 1.9139 |
| 5 | 2 | 1.4142 | 0.3010 | 86.26 ± 1.04 | 1.1481 | 1.9358 |
| 6 | 3 | 1.732 | 0.4771 | 93.56 ± 0.99 | 0.8089 | 1.9711 |
| 7 | 4 | 2 | 0.6020 | 99.14 ± 0.86 | -0.0655 | 1.9962 |

TABLE5:INVITRORELEASECHARACTERISTICS OFFORMULATIONS

| Time(h) | Cumulative%drugrelease | | | | | | | | |
|---------|------------------------|-------------|------------|------------|------------|------------|------------|------------|------------|
| | F1 | F2 | F3 | F4 | F5 | F6 | F7 | F8 | F9 |
| 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| 0.5 | 47.46±1.59 | 36.98±2.00 | 44.59±3.05 | 45.89±0.94 | 30.74± | 47.71±1.28 | 40.16±0.96 | 32.6±1.91 | 32.6±1.91 |
| 1 | 51.68±1.52 | 45.55±2.73 | 52.33±2.29 | 57.21±2.17 | 43.50±2.96 | 58.12±1.04 | 44.84±0.91 | 38.78±0.99 | 40.09±0.15 |
| 1.5 | 58.01±1.56 | 52.62±1.23 | 57.47±0.57 | 59.51±3.02 | 46.37±2.60 | 61.28±1.52 | 49.91±2.28 | 44.76±0.76 | 45.13±0.79 |
| 2 | 60.36±2.25 | 57.15±3.35 | 59.89±1.46 | 63.16±1.98 | 48.34±1.05 | 63.94±3.49 | 52.77±2.56 | 47.47±0.37 | 47.47±0.37 |
| 3 | 68.36±1.88 | 61.1±3.42 | 61.59±1.04 | 66.49±2.06 | 51.01±3.13 | 66.39±0.98 | 55.92±0.62 | 51.15±1.03 | 51.15±1.03 |
| 4 | 71.6±1.84 | 65.65±3.51 | 63.81±2.59 | 68.27±3.01 | 53.14±2.69 | 68.37±1.84 | 58.67±0.68 | 55.13±0.59 | 55.13±0.59 |
| 5 | 77.65±0.27 | 69.07±1.47 | 66.83±2.58 | 71.97±1.81 | 54.86±0.60 | 71.71±1.29 | 60.89±1.07 | 57.92±0.62 | 57.92±0.62 |
| 6 | 78.48±0.84 | 70.28±0.91 | 69.71±0.32 | 75.58±1.14 | 56.66±2.42 | 73.50±0.98 | 63.78±0.96 | 60.44±1.51 | 58.87±0.51 |
| 7 | 83.05±3.47 | 73.09±1.082 | 72.52±1.08 | 78.78±0.37 | 58.96±1.19 | 75±0.24 | 66.59±0.27 | 63.47±3.21 | 60.66±0.57 |
| 8 | 87.51±1.14 | 74.33±0.04 | 74.76±1.90 | 80.74±4.11 | 59.85±2.15 | 76.41±2.89 | 68.52±0.96 | 66.00±1.16 | 62.40±1.87 |

Formulations containing GG; F1, F2 and F3 released 87.51%, 74.33 and 74.76% of drug, formulations containing XG; F4 and F5 released 80.74% and 59.85% of the drug, formulations containing KG; F6, F7 and F8 released 76.41%, 68.52% and 66.00% of the drug and the formulations containing GG, XG and KG; F9 released 62.40% of the drug at the end of 8h respectively. Hence on the basis of *invitro* dissolution studies formulation F1 was chosen as the best formulation giving 87.51% of drug release till 8 h (Table 5).

CONCLUSION: In order to increase the local action in the gastrointestinal area against *Helicobacter pylori*, the current research set out to establish a delivery mechanism that might enable the retention of ofloxacin. In light of this, the current study aims to solve the problem of increasing the drug's bioavailability after oral administration by creating zinc pectinate beads entrapped in rice bran oil that contain ofloxacin.

Ofloxacin floating beads in a variety of formulations were created by combining LMP with rate-controlling polymers including GG, KG, and XG, or by employing LMP alone. The beads were made using the emulsion gelation technique. The beads made using a combination of LMP and GG (F1) exhibited the maximum drug release compared to other formulations. Rice bran oil was used to add buoyancy to the beads because of its low density. During the two-month stability investigation, the chosen formulation exhibited no further changes in drug content, floatability, or in vitro drug release profile while stored at $75 \pm 5\%$ RH at $40 \pm 2^\circ\text{C}$. So, the goal of this study was successfully accomplished: to create a dosage form of Ofloxacin utilizing a low density oil and various combinations of polymers that regulate the release rate.

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